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Solubility Rules. A  
solute is considered  
soluble if an  
appreciable amount of  
it can be dissolved in a  
given amount of the  
solvent. For example,  
both table salt ( $\text{NaCl}$ )  
and table sugar (C 11

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H<sub>2</sub>O) are soluble substances in water. A solute is considered insoluble if very little of it dissolves in a given amount of the solvent. For example, sand (SiO<sub>2</sub>) is considered insoluble in water.

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Answers to Solubility Rules Worksheet 3. Classify each of the substances as being



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soluble or insoluble in water.

- a. potassium bromide – sol
- b. lead (II) carbonate – insol
- c. zinc hydroxide – insol
- d. sodium acetate – sol
- e. silver iodide – insol
- f. zinc carbonate – insol
- g. silver acetate – insol
- h. copper (II) sulfide – insol
- i.

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ID: 1368034 Language:  
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Science Grade/level: 8

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To answer this, find where the curve of KCl crosses 30°C. I have marked this point A on the graph. You then read across and determine the grams of solute which is 37 g.

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Worksheet. For most  
substances, solubility

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increases as  
temperature increases.  
What are the  
exceptions on the  
graph below? \_\_\_\_\_

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worksheet, your child  
learns about differing  
rates of solubility by  
reading a bar chart and  
interpreting the data to  
answer questions. Not  
all soluble substances

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dissolve equally. In this  
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Worksheet: Solubility  
Graphs Name \_\_\_\_\_  
CHEMISTRY: A Study of  
... KEY Use the  
provided solubility  
graph to answer the  
following questions:  
For questions 1 - 4 an  
amount of solute is  
given, and a ... For  
questions 11 - 12,

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show your work and  
circle your final

answer. 11. If 115 g  
KNO<sub>3</sub> are added to  
100 g of water at 35  
°C, how many ...

**Use the provided  
solubility graph to  
answer the following**

...

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Use the solubility curve below to help you answer the following questions: You'll notice that for most substances, solubility increases as temperature increases. As discussed earlier in

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solutions involving liquids and solids typically more solute can be dissolved at higher temperatures.

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WORKSHEET KEY Use  
your solubility curve  
graphs provided to

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Worksheet And  
Answers

answer the following questions. 1. What are the customary units of solubility on solubility curves? Degree Celsius and grams of solute/100g of water 2. Define solubility. A measure of how much solute can dissolve in a given amount of solvent. 3.

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Some of the worksheets below are Predicting Precipitation Reactions Worksheets, use the solubility rules to identify which reaction would form a precipitate, learn the formation of an insoluble product (precipitate) after mixing of two electrolyte solutions, several solved exercises.

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About This Quiz & Worksheet. The quiz is a collection of straightforward questions. The questions will ask you about the characteristics and definitions of terms related to solubility and ...

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