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amount.... an equation in which each side has the same number of atoms of.... a concise representation of a chemical reaction using chemical....

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FFeagles. Terms in this set (15) activity series. a list of elements organized according to the ease with which the elements undergo certain chemical reactions. chemical equation.

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ANSWER Answer the following questions
in the space provided.

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equation. skeleton equation. balanced equation. precipitate. a concise representation of a chemical reaction using chemical.... a chemical equation that does not indicate the relative amount.... an equation in which each side has the same number of atoms of....

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number that appears in front of a formula in ...

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CHAPTER 8 REVIEW Class SĖc7ĠB-n
Chemical Equations and Reactions
SHORT ANSWER Answer the following
questions in the space provided. 1.

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Match the symbol on the left with its appropriate description on the right. (aq)
(a) (b) (c) (d) (f) A precipitate forms. A gas forms. A reversible reaction occurs. Heat is applied to the reactants.

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Chapter: Chemical Equations and

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Answers

Reactions PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. _____ 1. The production of a slightly soluble solid compound in a double-displacement reaction results in the formation of a a. gas. b ...

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Answers

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Modern Chemistry 71 Chapter Test

Name Class Date Chapter Test B,

continued 17. A substance combines with oxygen, releasing a large amount of energy as heat and light, in a(n) . 18.

The decomposition of a substance by an electric current is called . 19. A(n) orders

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Answers

the elements by the ease with which they undergo certain chemical reactions.
20.

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CHAPTER. 6. Stereochemistry.

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Intent and Purpose Stereochemistry is the.

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6723 Chapter 8 Test Page 6 19 Your teacher gives you the problem $5 \div 1 \underline{\quad} 4$
Part A ... Holt McDougal Modern Chemistry Chapter Test Assessment

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Chapter Test B Teacher Notes and
Answers 5 The Periodic Law TEST B 1 a 2
c 3 d 4 d 5 a 6 a 7 c 8 a 9

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Name Class Date Chapter Test B,
continued 18. When a strong acid is
titrated with a weak base, the pH of the

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solution at the equivalence point is than
7. 19. A is a highly purified solid used to
check the concentration of a standard
solution. 20. A 1 M solution of NaOH will
have a pH that is

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Name _____ Class _____ Date _____ Chapter Test A,
continued _____ 21. Metal X replaces the
ions of metal Y from solution, but it
cannot replace the ions of metal Z from
solution. The order these metals should
have in the activity series (from top to
bottom) is a. x, y, z b. z, x, y c. y, x, z d.
z, y, x _____ 22.

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Modern Chemistry 1 Chemical Equations
and Reactions CHAPTER 8 REVIEW

Chemical Equations and Reactions

Teacher Notes and Answers Chapter 8

SECTION 1 SHORT ANSWER 1. a. d b. a

c. b d. f e. e f. c 2. 8,4,9 3.

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Answers

Modern Chemistry Chapter 8 Review Answers Section 2

Chapter 1 an Introduction to Chemistry
3 I would watch the buds swell in spring,
the mica glint in the granite, my own
hands, and I would say to myself: "I will
understand this, too Modern chemistry
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