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## **Compounds Their Formulas Lab 7**

Compounds and Their Formulas Report  
Sheet Lab - 7 B.1 Physical properties

Compound Appearance Density Melting  
Point B.2 Formulas of Ionic compounds

Name Positive Ion Negative Ion Formula

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Sodium chloride Na + Cl-Magnesium  
chloride Calcium oxide Lithium  
phosphide Aluminum sulfide Calcium  
nitride B.3 Names of Ionic compounds K  
2 S Potassium sulfide BaF 2 MgO Na 3 N  
AlCl 3 Mg 3 P 2 C. Ionic Compounds with  
Transition Metals C.1 Physical properties

## **lab 7 - Compounds and Their**

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## **Formulas Report Sheet Lab 7 ...**

Compounds and Their Formulas Report Sheet - Lab 7 Date Section Instructor Name Team Pre-Lab Study Questions 1. Where are the valence electrons in an atom? 2. How are positive and negative ions formed? 3. How do subscripts represent the charge balance of ions? 4. Why are electrons shared in covalent

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compounds? 5.

## **Solved: Compounds And Their Formulas Report Sheet - Lab 7 ...**

Compounds and Their Formulas Report  
Sheet Lab - 7 E.3 Electron-dot structures  
Compound Electron-Dot Structure Name  
H<sub>2</sub>O SBr<sub>2</sub> PCl<sub>3</sub> CBr<sub>4</sub> SO<sub>3</sub> Questions  
and Problems Q.2 a. Identify each

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following compounds as ionic or covalent. b. Write the correct formula for each.

## **Compounds and Their Formulas Report Sheet Lab 7 E3 ...**

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Compounds and Their Formulas Pre-lab



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Study Questions 1. Where are the valence electrons in an atom? In the last shell of an

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have a lot of extra features that make it a go-to place when you're looking for free Kindle books. Compounds Their Formulas Lab 7 For many formulas, however, more than one compound is possible. For

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Laboratory 7 Report Sheet - Lab 7 C.  
Ionic Compounds with Transition Metals  
C.1 Physical properties Compound  
Appear C.2 Formulas of ionic compounds  
Density Melting Point Negative Ion  
Formula  $FeCl_3$  Name Iron(III)  
chloride Iron(II) oxide  
Copper sulfide  
C.3 Names of ionic compounds  
Copper

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Chemical Compound Formulas.  
Chemistry is a science of the  
composition of matter and  
transformation of its physical or  
chemical property. Our universe is made

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of matter. The matter is composed of atoms of various different chemical elements that are found in pure as well as combined form. Any substance consisting of the matter is called chemical.

## **Chemical Compound Formula | Formula Chart**

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In order to write a chemical formula, it is important to know the symbol of the elements present in the compound, formula of the radicals and the valency of the elements in that compound. Following points should be kept in mind while writing a chemical formula. Most of the compounds are binary compounds, i.e. they have two elements.



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## **Chemical Formula of Common Compounds, Chemical reaction ...**

For many formulas, however, more than one compound is possible. For example, the CRC Handbook lists two compounds having the formula  $C_2H_6O$ , three compounds having the formula  $C_2H_2Cl_2$ , and twenty-one compounds having the

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formula  $C_4H_{10}O_2$ . Each compound is a different substance, with its own unique set of physical and chemical properties.

### **Lab #7**

Compounds and Their Formulas Pre-lab Study Questions 1. Where are the valence electrons in an atom? In the outermost shell occupied in an atom 2.

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How specifically are positive and negative ions formed? Positive ions are formed when one atom loses one or more electron to another, ...

## **LAB E DONE - Compounds and Their Formulas Pre-lab Study ...**

7. Cd 3P2 Cd 2+, cadmium ion P3-, phosphide ion Cadmium phosphide

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Molecular Compounds (Non-Metal + Non-Metal) Compound Formula Compound Name  
1. ICl Iodine monochloride  
2. N<sub>2</sub>O<sub>5</sub> Dinitrogen pentoxide  
3. BrF<sub>5</sub> Bromine pentafluoride  
4. C<sub>3</sub>H<sub>8</sub> Tricarbon octahydride

### **Laboratory #6: Naming Compounds**

Report Sheet-Lab 7 D.2 Formulas of ionic

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compounds Formula Positive Ion  
Negative Ion Name Potassium carbonate  
K Sodium nitrate Calcium bicarbonate  
Aluminum hydroxide Lithium phosphate  
Potassium sulfate Co,2- D.3 Names of  
ionic compounds CaSO<sub>4</sub> AlNO<sub>3</sub> Na CO<sub>3</sub>  
MgSO<sub>4</sub> Calcium sulfate Mg (PO<sub>4</sub>)<sub>2</sub>  
Questions and Problems Q.1 Write the  
correct formulas for the following ions:

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sodium ion oxide ion calcium ion ...

## **Solved: Report Sheet-Lab 7 D.2 Formulas Of Ionic Compounds ...**

Butane has the empirical formula  $C_2H_5$ , but it contains two  $C_2H_5$  formula units, giving a molecular formula of  $C_4H_{10}$ . Because ionic compounds do not contain discrete molecules, empirical

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formulas are used to indicate their compositions. All compounds, whether ionic or covalent, must be electrically neutral.

## **3.1: Types of Chemical Compounds and their Formulas ...**

Ionic And Covalent Compounds Lab 3  
Lab Instructions. WEEK 3, LAB 3 OUT OF

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