

# Chapter 2 Chemistry Of Life

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Biology Chapter 2- The Chemistry of Life Essential Question: What are the basic chemical principles that affect living things? 2.1 The Nature of Matter What 3 subatomic particles make up atoms?

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Chapter 2: The Chemistry of Life. Chapter 2 Vocabulary. Atom Nucleus Electron Element Isotope Compound Ionic bond Ion Covalent bond Molecule. Van der Waals forces Hydrogen bond Cohesion Adhesion Mixture Solution Solute Solvent Suspension pH scale. Acid Base Buffer Monomer Polymer Carbohydrate Monosaccharide Lipid Nucleic acid Nucleotide.

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2.3 Carbon Compounds Lesson Objectives Describe the unique qualities of carbon. Describe the structures and functions of each of the four groups of macromolecules. Lesson Summary The Chemistry of Carbon Organic chemistry is the study of compounds with bonds between carbon atoms. Carbon atoms have four valence electrons, allowing them to form

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A B; atom: the basic unit of matter: nucleus: the center of the atom: electron: a negatively charged ...

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CHAPTER OUTLINE 2.1: The Building Blocks of Molecules 2.2: Water 2.3: Biological Molecules Foods such as bread, fruit, and cheese are rich sources of biological macromolecules. KEY BUILDING BLOCKS OF LIVING ORGANISMS Key elements :  
•Carbon (C) •Hydrogen (H) •Nitrogen (N) •Oxygen (O) •Sulfur (S) •Phosphorus (P) Form : •Carbohydrates •Nucleic acids •Proteins •Lipids

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