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Modern Chemistry Chapter 3 Atoms: The Building Blocks of Matter law of conservation of mass- mass is neither created nor destroyed during ordinary chemical reactions or physical changes e.g. $20\text{ g A} + 20\text{ g B} \rightarrow 40\text{ g AB}$

Modern Chemistry Chapter 3 Atoms: The Building Blocks of ...

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3 Atoms: The Building Blocks of Matter

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Title: Modern Chemistry Chapter 3 Atoms: The Building Blocks of Matter 1 Modern Chemistry Chapter 3 Atoms The Building Blocks of Matter. law of conservation of mass- mass is neither created nor destroyed during ordinary chemical reactions or physical changes ; e.g. 20 g A + 20 g B → 40 g AB; 2. law of definite proportions- a chemical compound

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MODERN CHEMISTRY The atomic number of nickel-60 is 28. How many neutrons does this iso- tope have? Carbon-14 has 8 neutrons. What is the atomic number of carbon-14? 11 CHAPTER 3 TEST HR W material copyrighted under notice appearing earlier in this work.

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Holt McDougal Modern Chemistry 3 Chapter Test Chapter Test B, continued 16 Modern chemistry chapter 3 test b answers. The measure of the ability of an atom in a chemical compound to attract electrons from another atom in the compound is called _____. 17. The energy required to remove one electron from an atom is called its _____. 18.

Modern Chemistry Chapter 3 Test B Answers

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Modern Chemistry Chapter 3 Section Answers

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